

These notes can be obtained at: <http://www.ndsu.nodak.edu/instruct/grcook/chem342/notes.shtml>

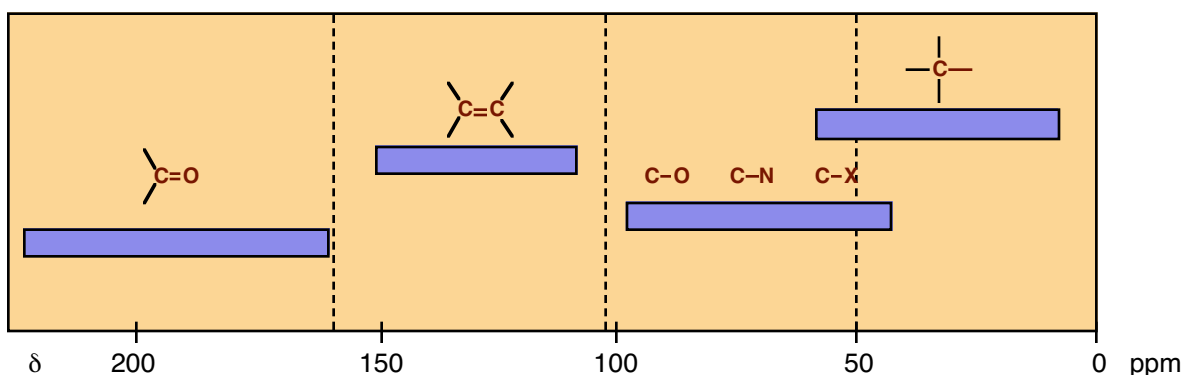
## Chapter 13: Nuclear Magnetic Resonance Spectroscopy

### <sup>13</sup>C NMR Spectroscopy

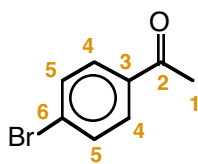
Gives the number of chemically different carbons in the molecule.

Most carbons show resonances in the range of 0 to 220 ppm. There are four regions which one can find different kind of carbons.

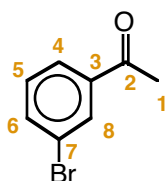
#### <sup>13</sup>C NMR



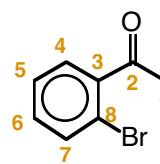
Carbons which are related by symmetry are chemically equivalent and show up as one peak.



**symmetrical**  
**6 peaks**



**not symmetrical**  
**8 peaks**



**not symmetrical**  
**8 peaks**

A DEPT spectra (Distortionless Enhancement by Polarization Transfer) can tell you how many different H's are attached to each carbon. Three spectra are displayed. One shows all the different carbons in the molecule, one shows only the carbons with ONE hydrogen attached. The third shows all carbons with an odd number of H's attached as a positive peak (CH and CH<sub>3</sub>) and carbons with an even number (CH<sub>2</sub>) as a negative peak.