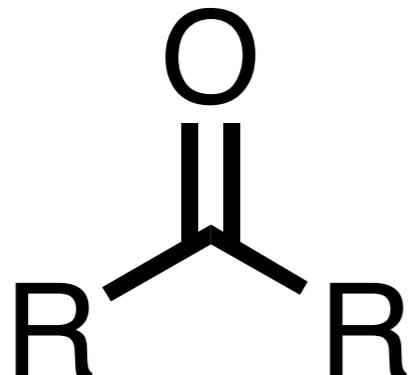


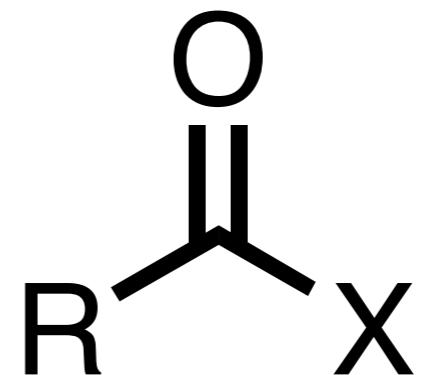
Final Exam
Friday, 12:30 pm
Stevens

~40-50% new material
~50-60% comprehensive

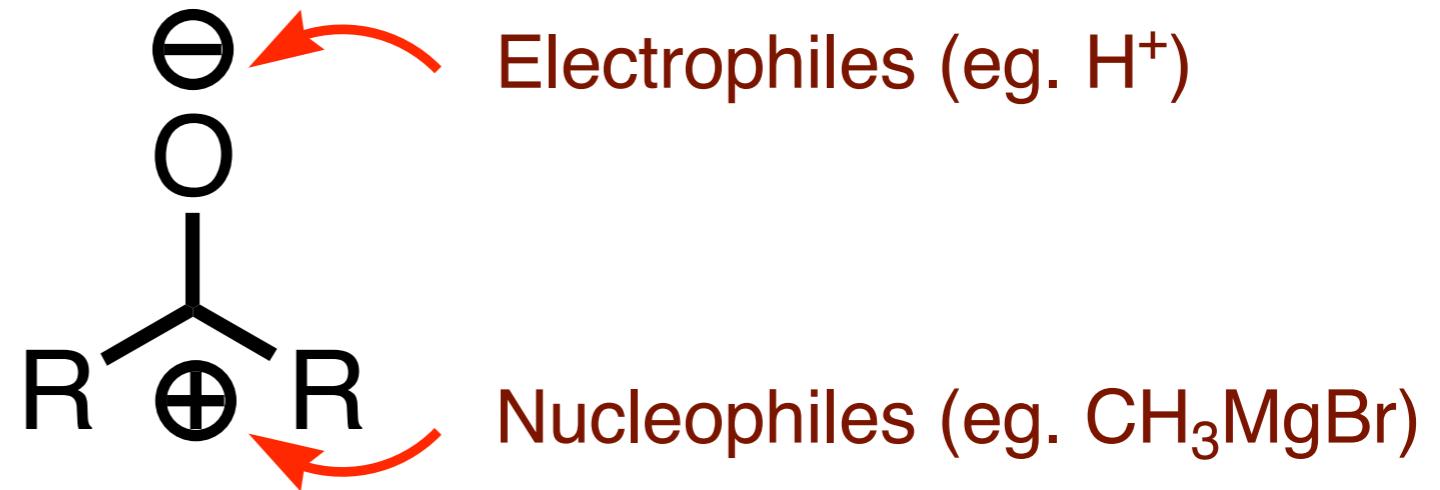
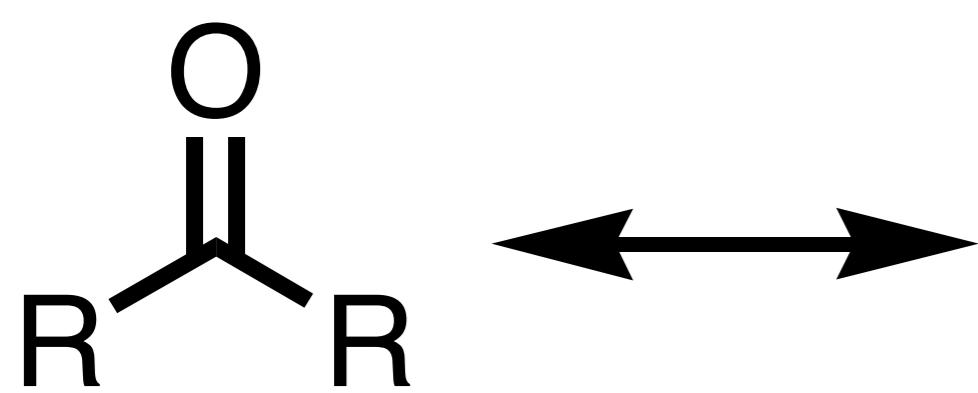
Carbonyl Chemistry



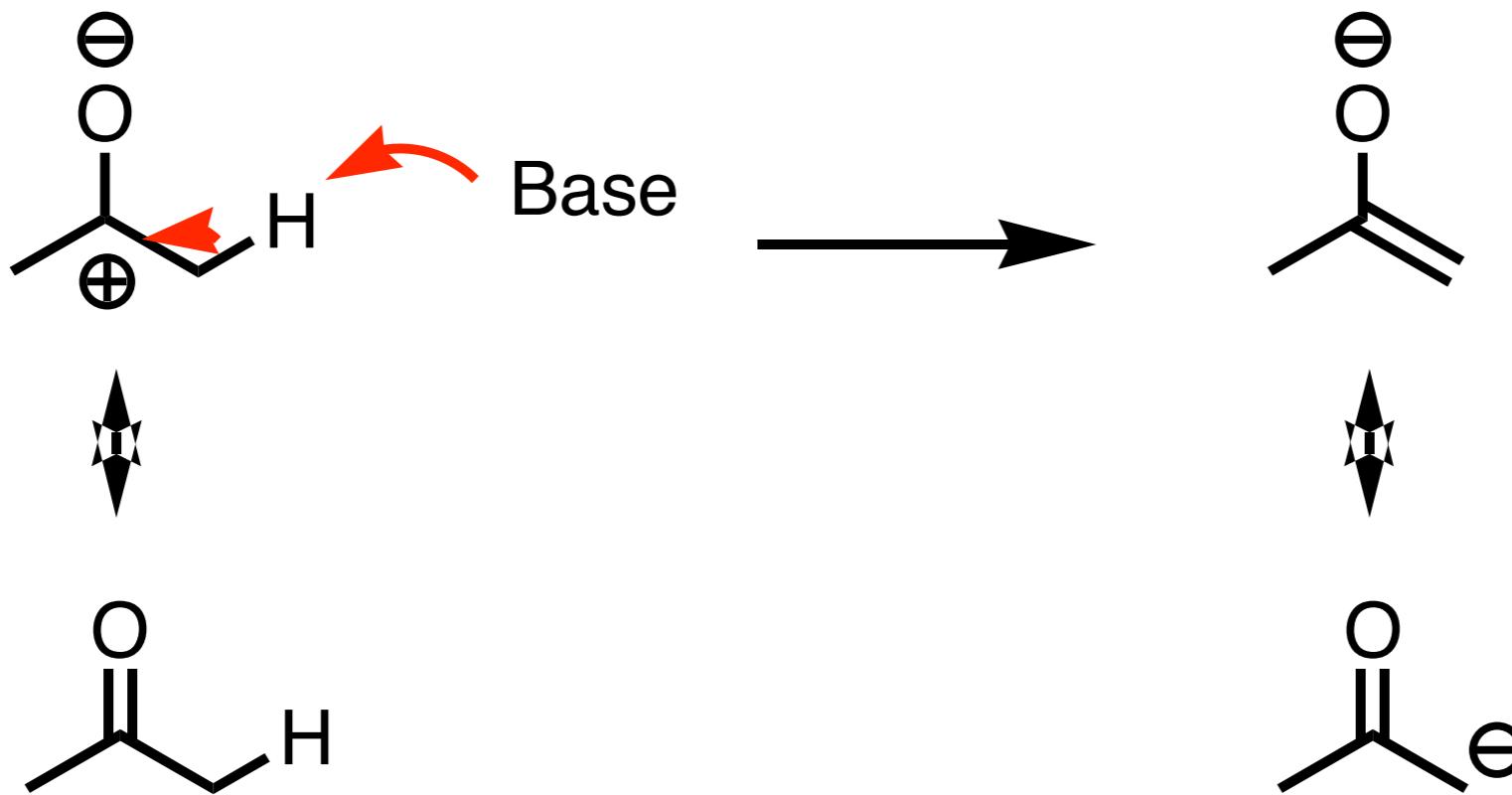
aldehydes
ketones

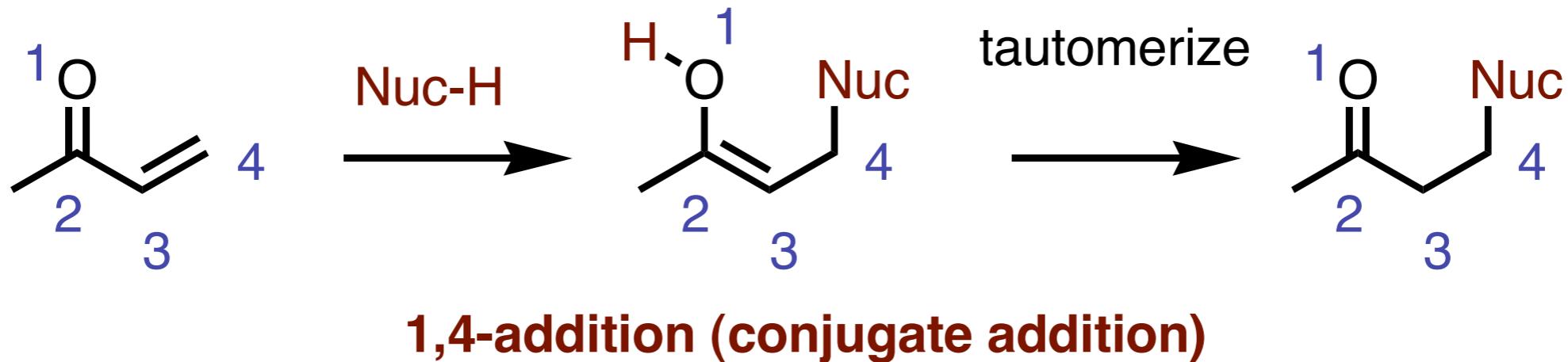
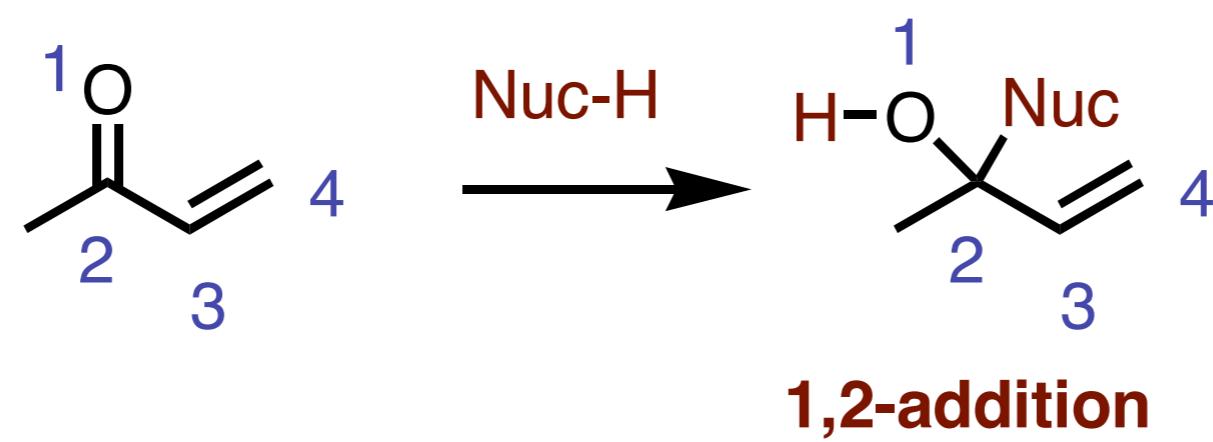
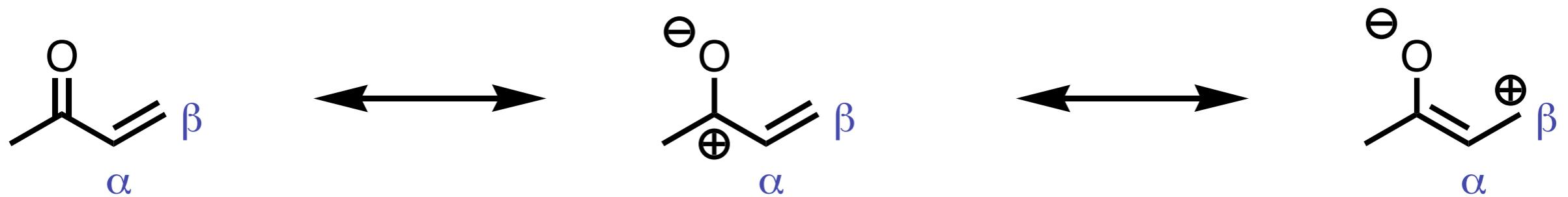


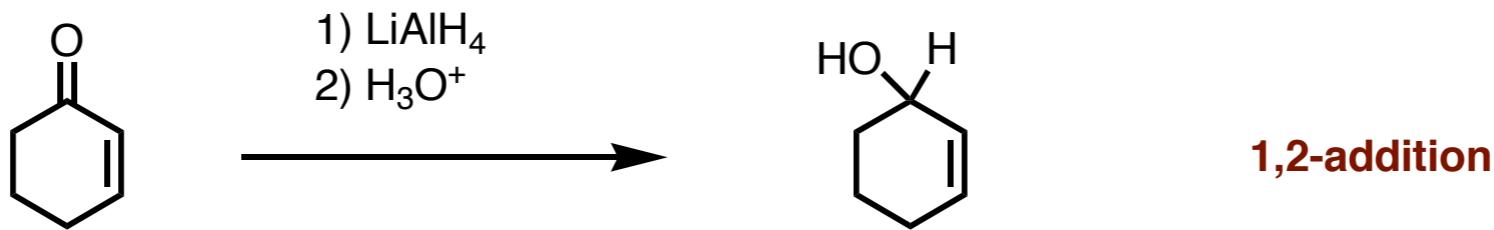
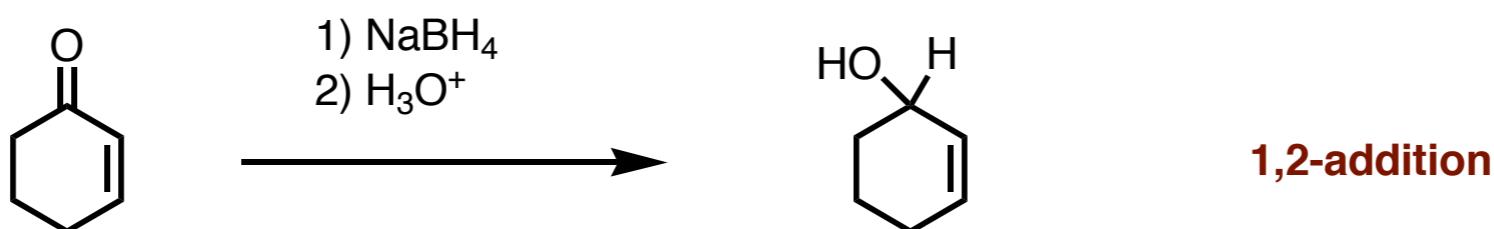
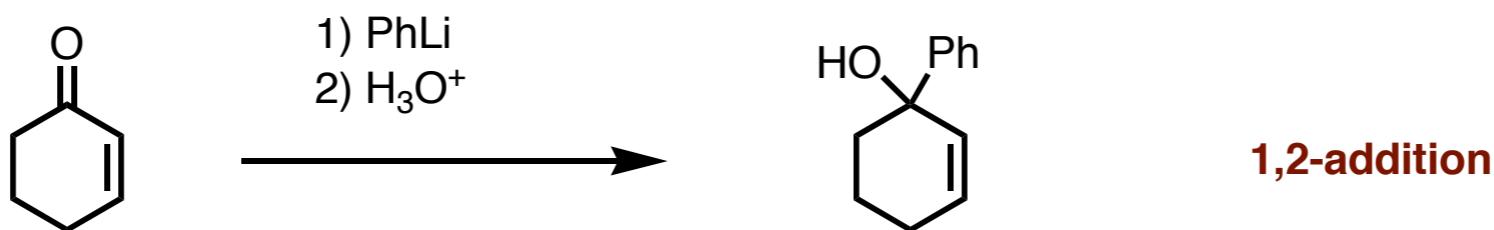
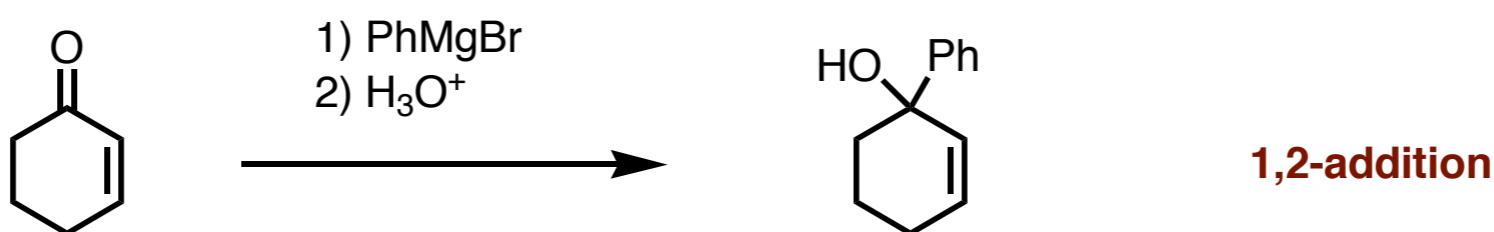
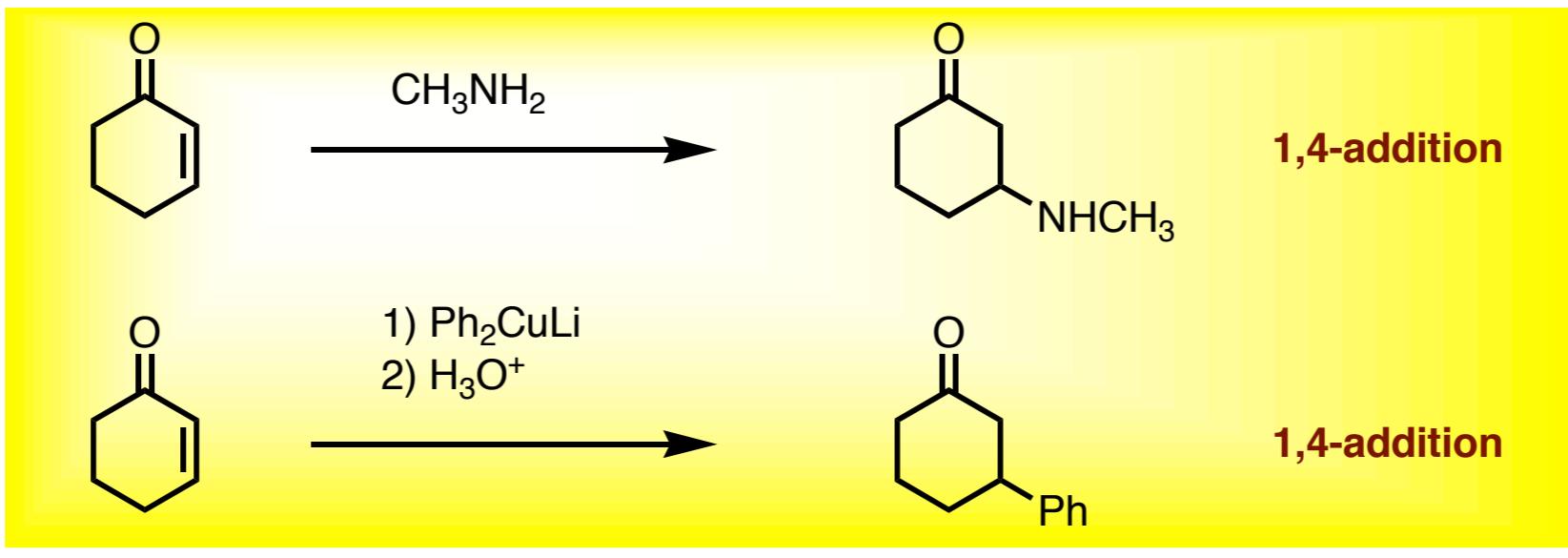
carboxylic acid
and derivatives

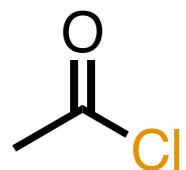
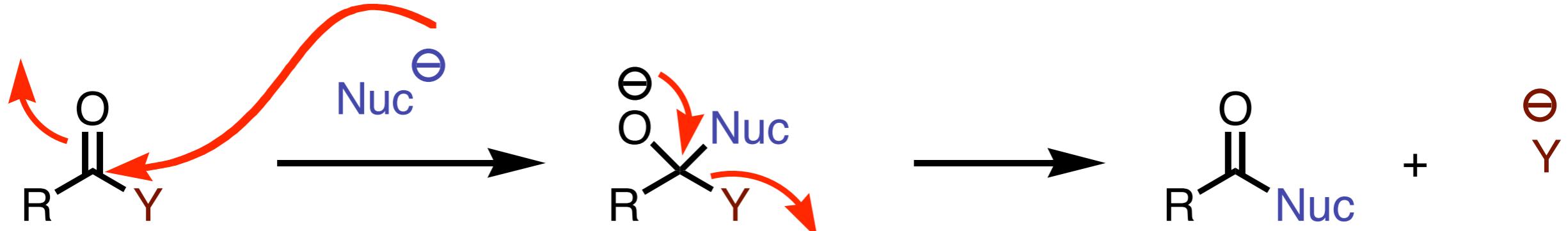


an enolate

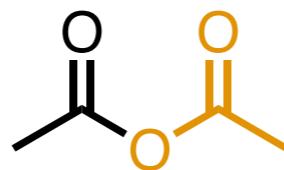








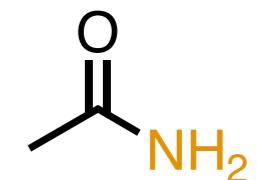
Acid Halides
(acetyl chloride)



Acid anhydrides
(acetic anhydride)



Esters
(methyl acetate)

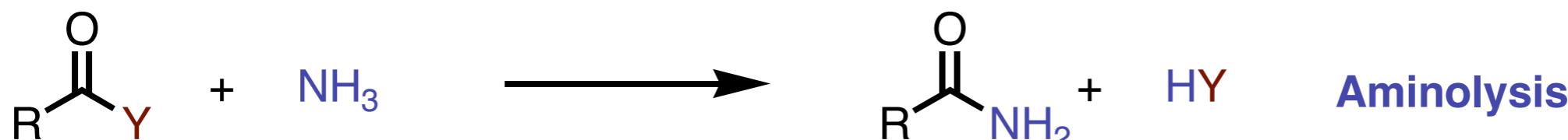
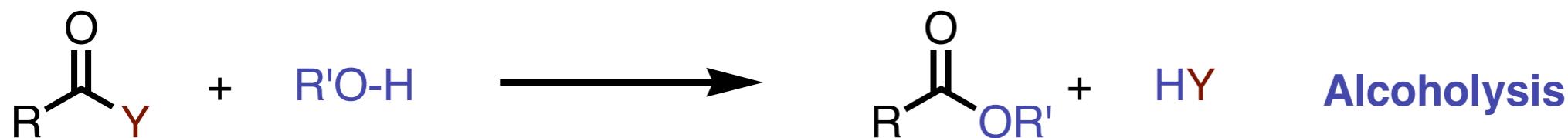
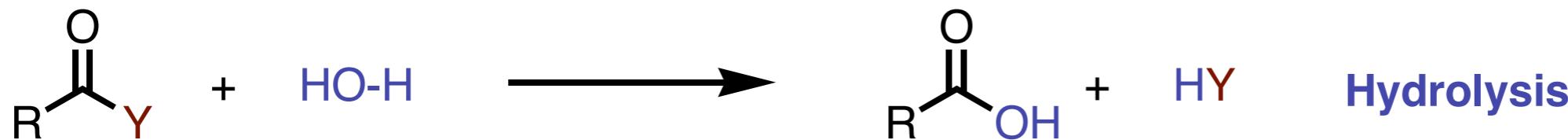
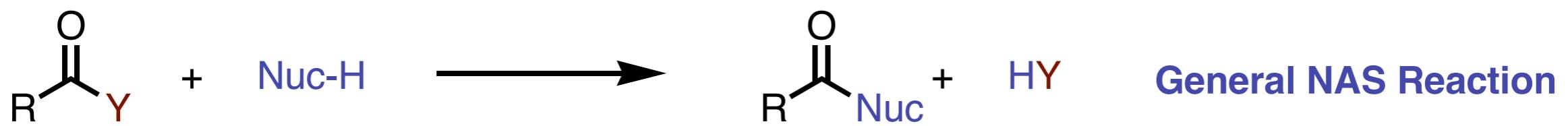


Amides
(acetamide)

More Reactive

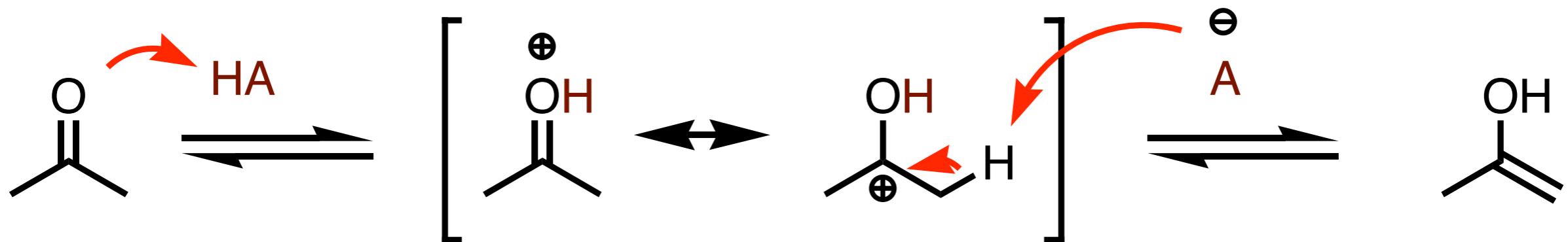


Less Reactive

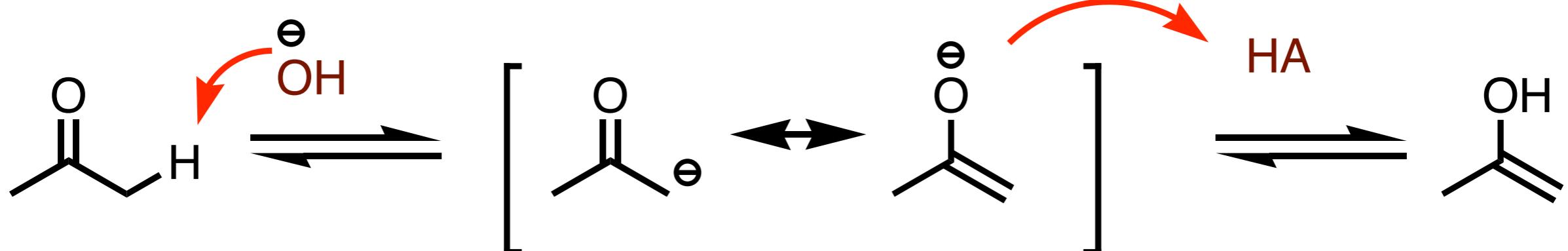


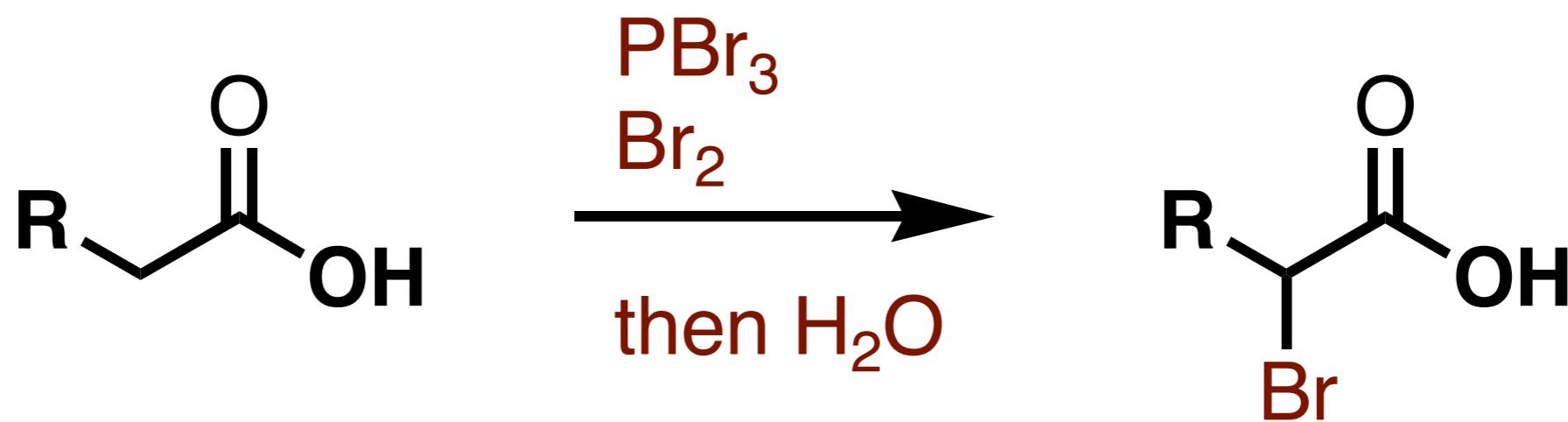
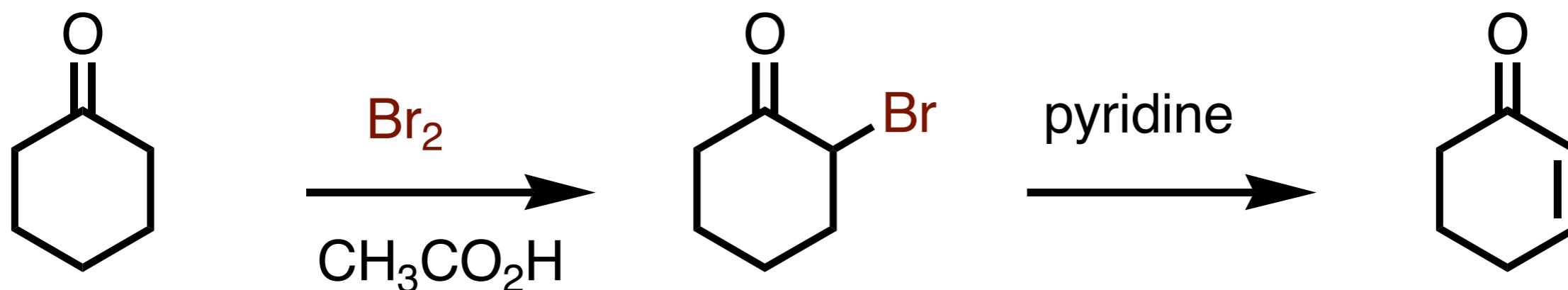
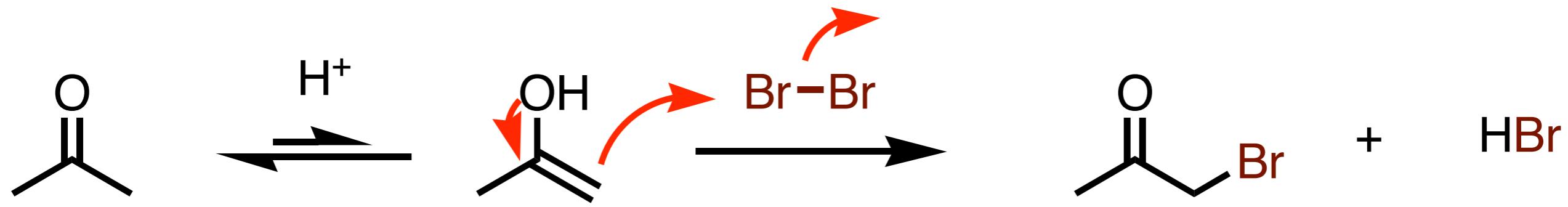


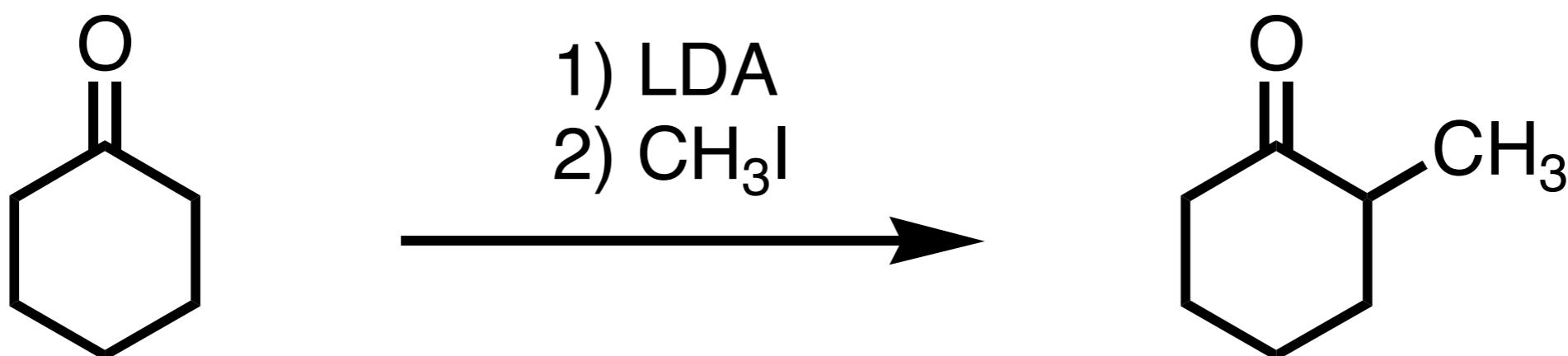
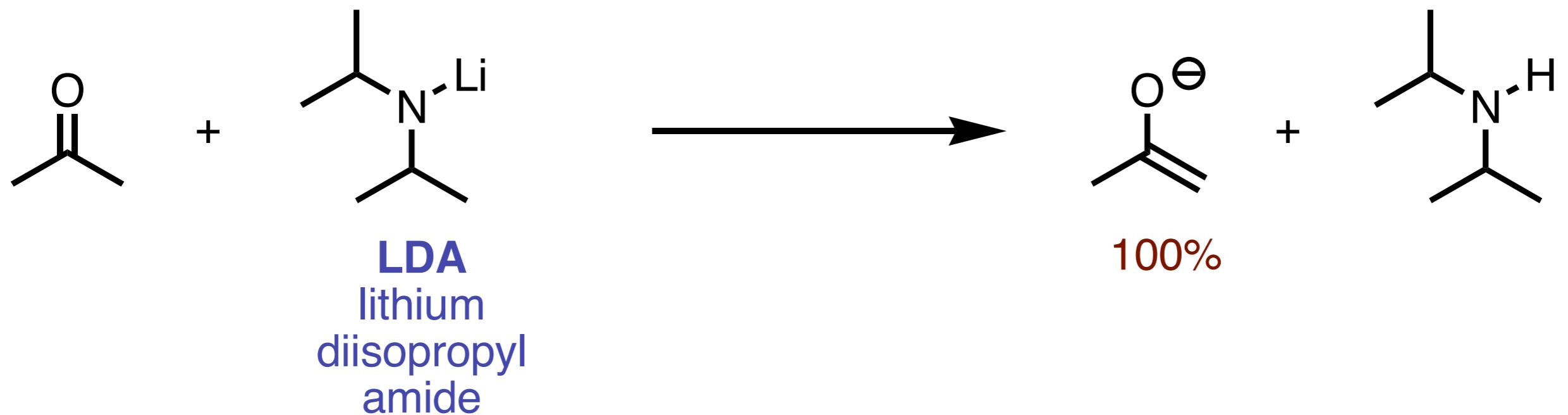
Acid Catalyzed Enol Equilibrium

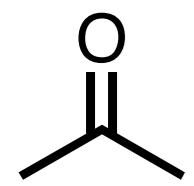


Base Catalyzed Enol Equilibrium

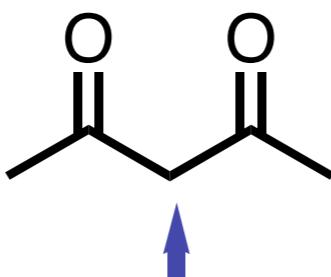




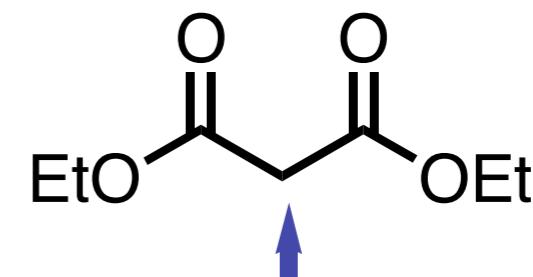




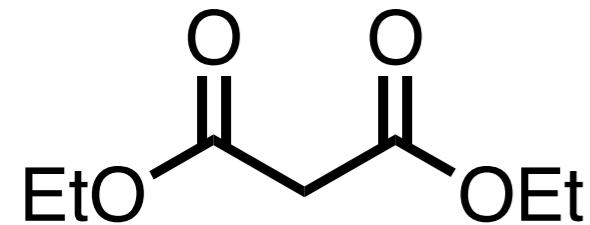
pK_a = 20



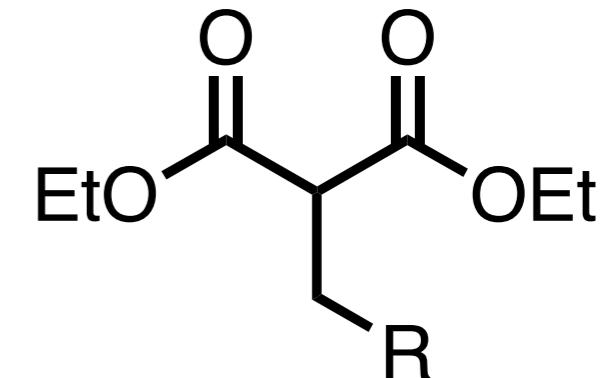
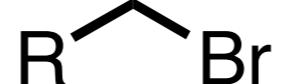
pK_a = 9

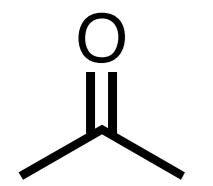


pK_a = 13

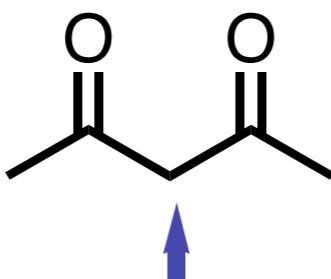


NaOEt, EtOH

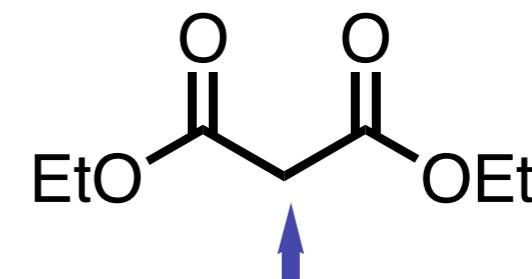




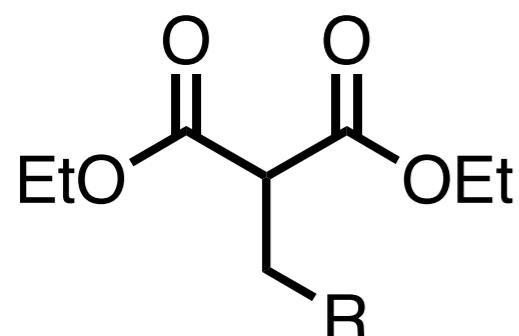
pK_a = 20



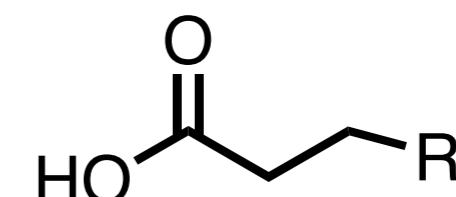
pK_a = 9



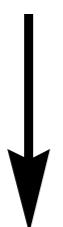
pK_a = 13



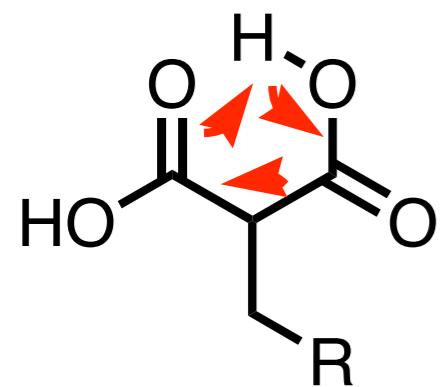
H₃O⁺, Heat



+ CO₂

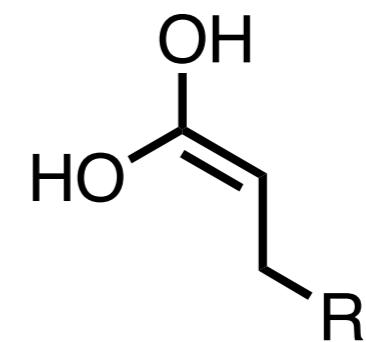


H₃O⁺

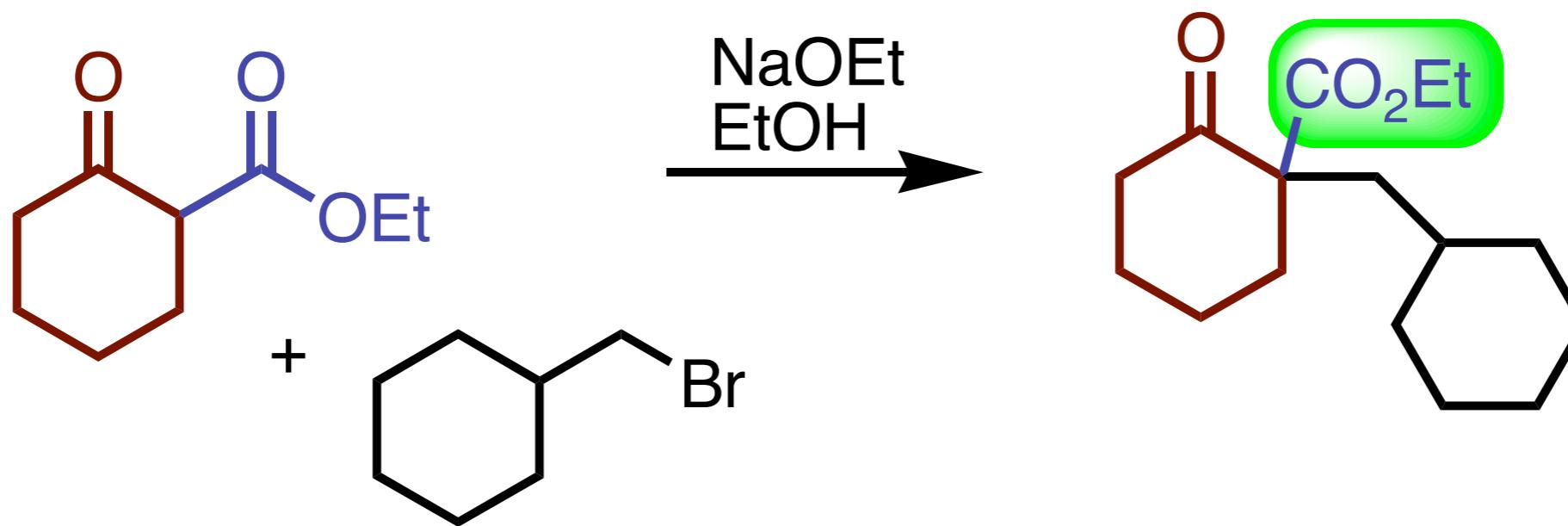


Heat

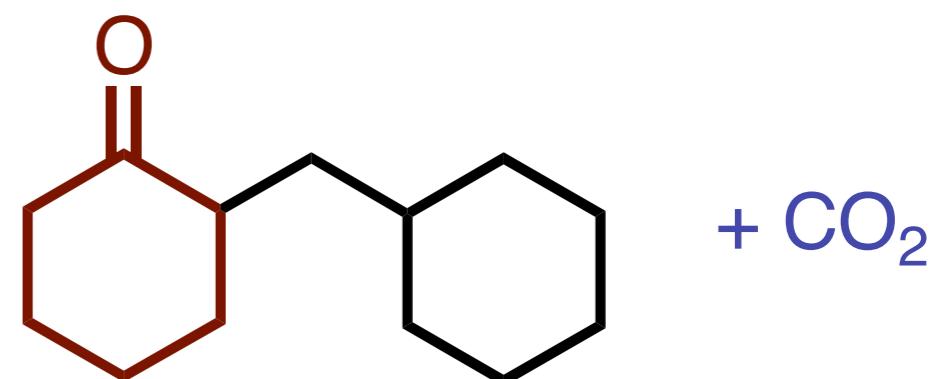
-CO₂

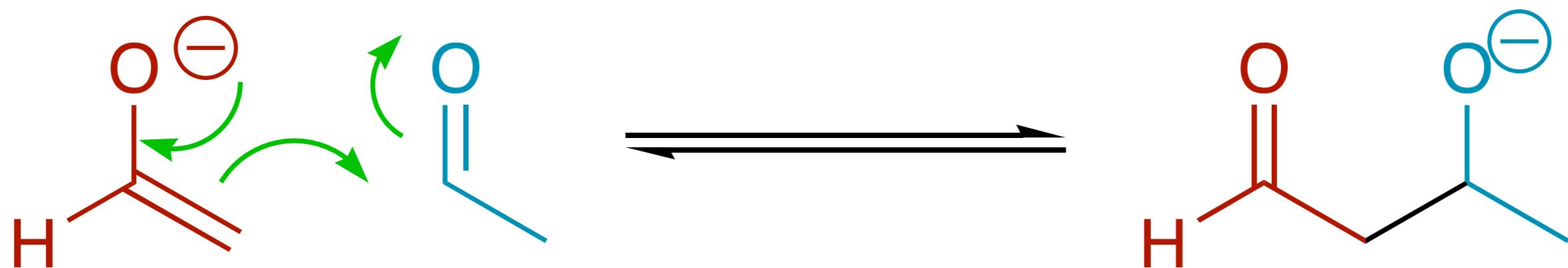
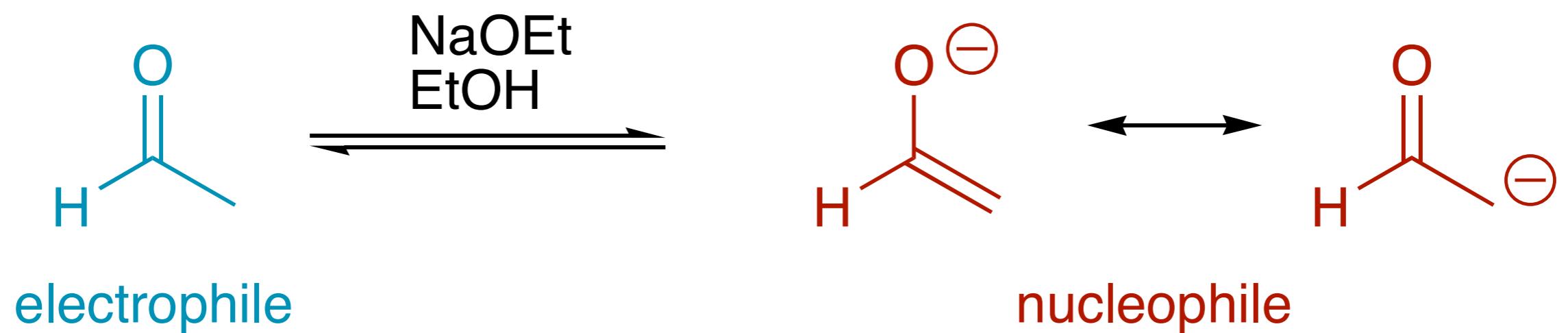


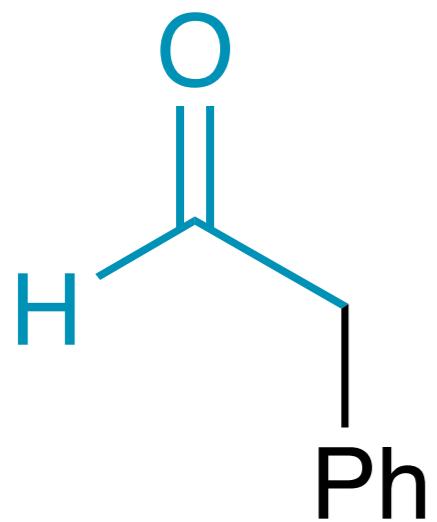
tautomerize



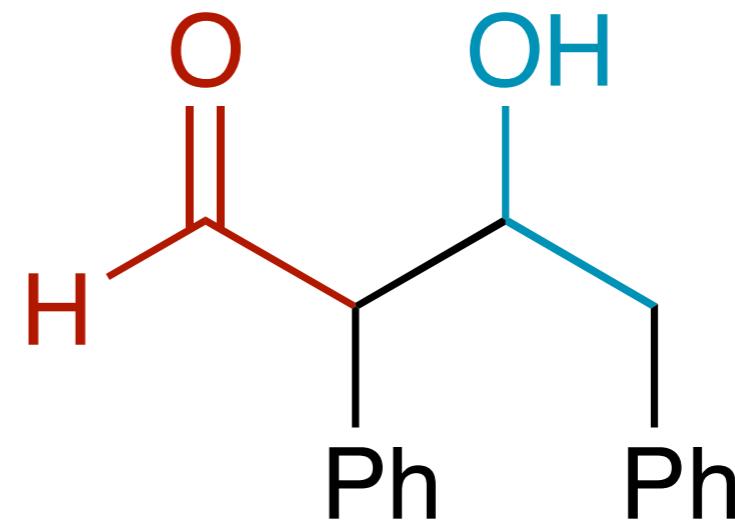
H_3O^+ , Heat



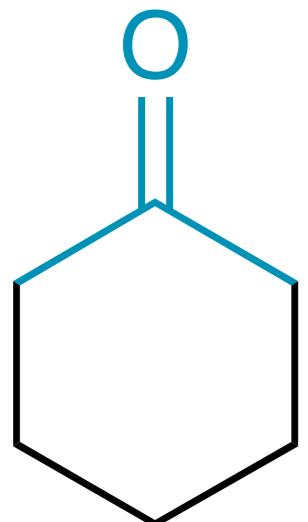




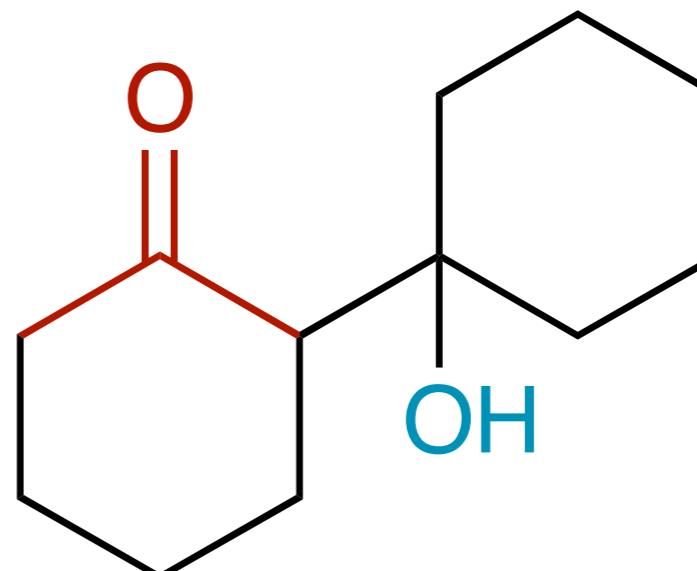
NaOEt
EtOH



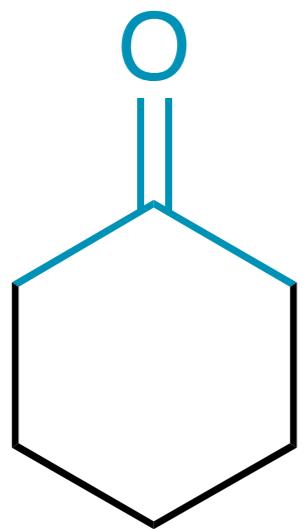
90%



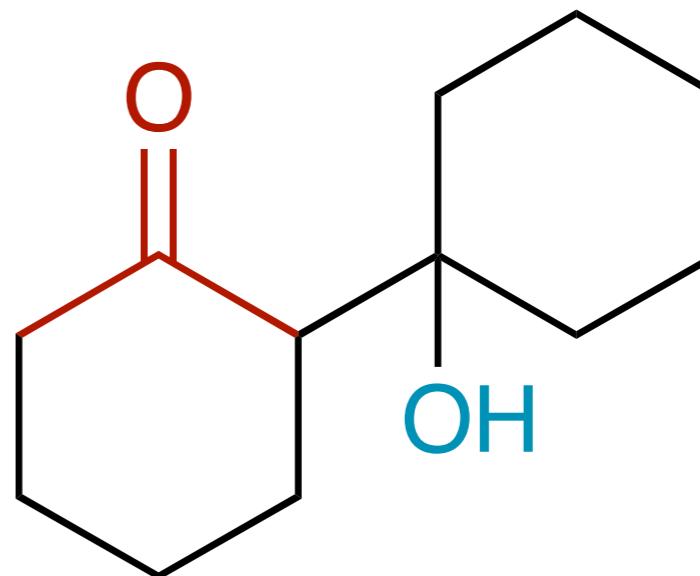
NaOEt
EtOH



22%

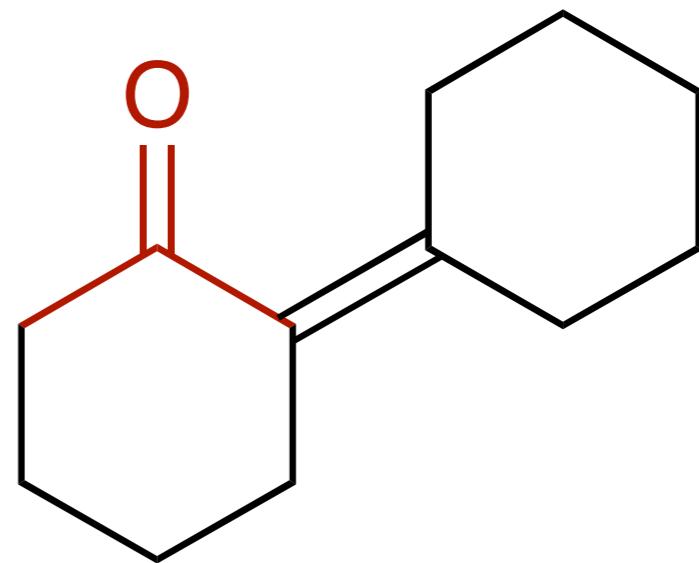


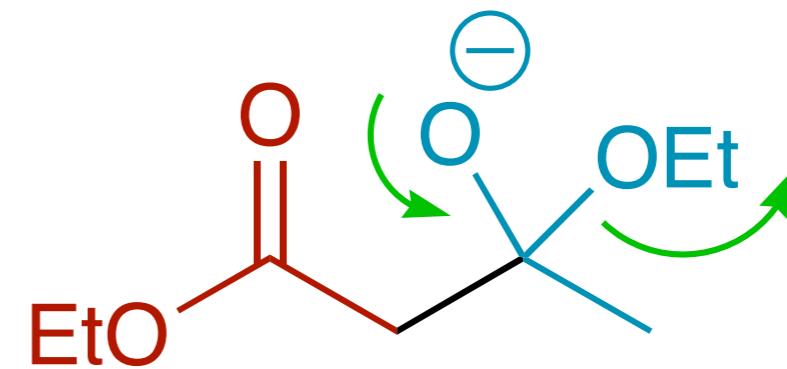
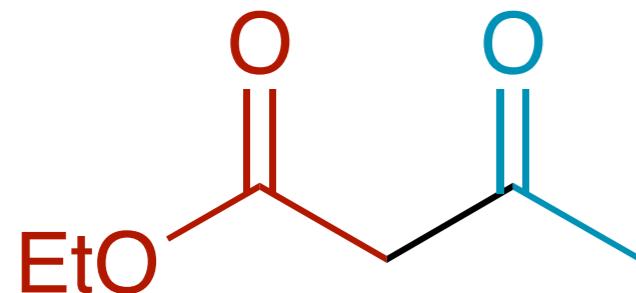
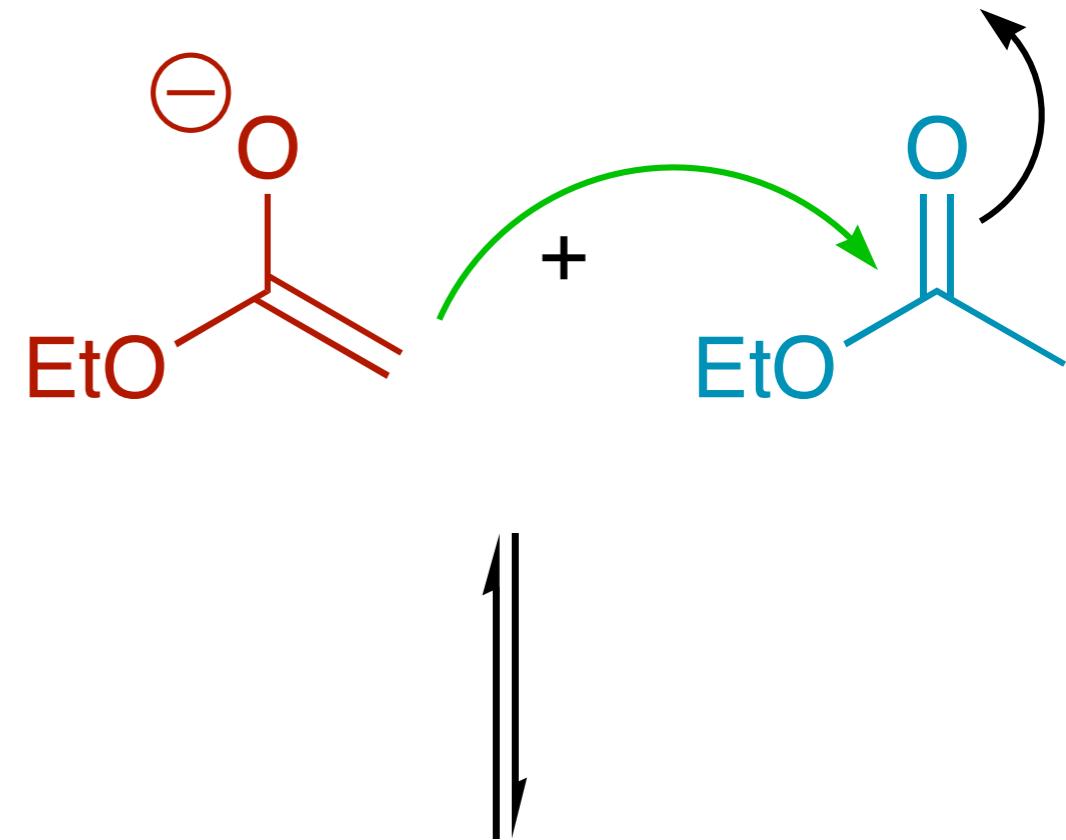
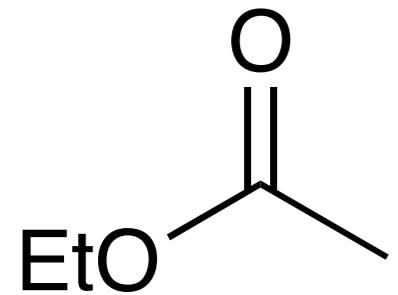
NaOEt
EtOH

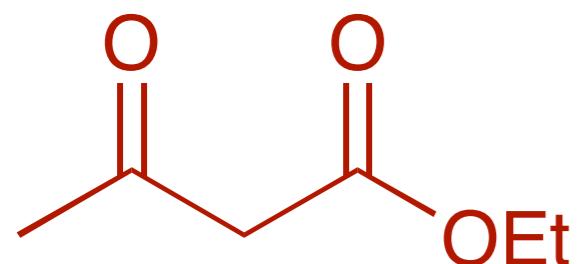


22%

heat
-H₂O







$\xrightleftharpoons[\text{EtOH}]{\text{NaOEt}}$

